**Task 3 Volumetric Analysis Practical Test 2018 Name: \_\_\_\_\_\_\_\_\_\_\_\_\_\_**

**Task description:** You are required to determine the concentration of ammonia (NH3) in the undiluted household cleaner. You can use any pieces of chemistry equipment in the laboratory.

**Equipment in tray:**

* Bromothymol blue – dropper bottle
* Methyl orange – dropper bottle
* Phenolphthalein – dropper bottle
* Plastic pipette – x2
* Plastic distilled water bottle

**Equipment in stock bottles at the front of the room:**

* 0.01M HCl solution – standardised solution
* Household cleaner solution – undiluted approximately 20 g L-1

**Initial planning section:**

1. Within the area below, with the assistance of labelled diagrams, explain how you are going to perform the activity. (5 marks)
2. Which indicator are you going to use? \_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_ (1 mark)
3. Explain why you chose that particular indicator? You may want to use a labelled pH titration graph to help. (2 marks)

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**Perform the experiment – check with your teacher first that your activity is safe**

1. **Results:** (3 marks)

**During the experiment**

1. What did you use to rinse the burette with? \_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_

Why? \_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_

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1. What did you use to rinse the pipette with? \_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_

Why? \_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_

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1. What did you use to rinse the conical flask (aliquot) with? \_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_

Why? \_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_

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1. Use a diagram to show how you accurately read the fluid level in the burette. (1 mark)

**Analysis of results:**

1. Write a balanced equation for the reaction between hydrochloric acid and ammonia

(2 marks)

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1. Determine the ammonia concentration in the undiluted household cleaner

(4 marks)

**Post activity analysis:**

1. Give two (2) system errors associated with you activity (1 mark)

11.1 \_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_

11.2 \_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_

1. How does the accuracy of the burette affect the results? (1 mark)

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1. If you happened to use the incorrect indicator, how could that have affected your overall result?

(2 marks)

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1. Give 3 ways in which you could improve your activity. Explain how they would improve the activity. (3 marks)
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1. Ammonia is a very effective cleaning agent and disinfectant. Why is the concentration of ammonia in household cleaner so low? (1 mark)

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1. If concentrated ammonia was spilt on the floor in the laboratory, what would be the safest way to have it cleaned up? Why?

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(2 marks)

**END OF TASK**